Hiroshi Maruoka, Motoyoshi Yamazaki and Yukihiko Tomioka*

Faculty of Pharmaceutical Sciences, Fukuoka University, 8-19-1 Nanakuma, Jonan-ku, Fukuoka 814-0180, Japan

Received December 10, 2001


#### Abstract

The reactions of 2-amino-4,5-dihydro-3-furancarbonitriles $\mathbf{1 a}$-d with $\alpha, \beta$-unsaturated carbonyl compounds in the presence of sodium ethoxide ( 0.1 equivalent) gave the corresponding Michael adducts $\mathbf{2 a - d}$, 3a-d and 4a-d. Compounds 2a-d and 3a-c reacted with sodium alkoxide (1 equivalent) to yield the corresponding 7a-alkoxyhexahydrofuro[2,3-b]pyridines 5a-d, 6a-d, 7a-c and 8a-c. Treatment of 5a-d, 6a-d, 7a-c and 8a-c with potassium tert-butoxide produced the corresponding dihydrofuro[2,3-b]pyridines $\mathbf{9 a - d}$ and 10a-c. The reaction of $\mathbf{4 a - c}$ with sodium ethoxide ( 1 equivalent) afforded the corresponding dihydro-furo[2,3-b]pyridines 11a-c.


J. Heterocyclic Chem., 39, 743 (2002).

The ir and ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectra of 2-amino-4,5-dihydro-3furancarbonitriles 1a-d [1] indicate that $\mathbf{1}$ exist as the primary enaminonitrile forms [2] rather than the imine forms. Therefore, depending on the experimental conditions used, electrophiles are expected to attack at either the enamino nitrogen atom or the $\beta$-carbon atom (the 3-position of $\mathbf{1}$ ) in the enamine. In fact, we showed that benzoyl chloride in the presence of sodium hydride attacks the 3-position of $\mathbf{1}$ to give the intermediate 3-benzoyl-2-imino-3-furancarbonitriles which undergo a ring opening to provide the malononitrile derivatives [3], whereas benzoyl chloride in pyridine attacks the amino group of $\mathbf{1}$ to yield 2-benza-mido-4,5-dihydro-3-furancarbonitriles [1]. When $\alpha, \beta$ unsaturated carbonyl compounds are used as electrophiles, conjugate addition of the $\beta$-carbon atom of enamines to $\alpha, \beta$-unsaturated carbonyl compounds takes place to produce the initial Michael adducts $[4,5]$. We now report the results of the reactions of $\mathbf{1}$ with $\alpha, \beta$-unsaturated carbonyl compounds in the presence of a base.

When a solution of $\mathbf{1 a - d}$ with methyl vinyl ketone and sodium ethoxide ( 0.1 equivalent) in diethyl ether or tetrahydrofuran was stirred at room temperature, the expected Michael adducts 2a-d were obtained. The conjugate addition of the amino group of $\mathbf{1}$ to methyl vinyl ketone was not observed. The ir spectra of 2a-d display a band at near $2230 \mathrm{~cm}^{-1}$ due to a non-conjugated cyano group, whereas those of 1a-d show a conjugated cyano band in the $2170 \sim 2190 \mathrm{~cm}^{-1}$ region. The ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectra of 2a-d exhibit a one-proton singlet at near $\delta 7.30$ attributable to an imino group, whereas those of $\mathbf{1 a - d}$ appear as a twoproton singlet at near $\delta 5$ assignable to an amino group. Compounds 1a-d reacted with benzalacetone and cinnamaldehyde under the same conditions to give the corresponding Michael adducts 3a-d and 4a-d. Elemental analyses and spectral data of $\mathbf{2 - 4}$ are consistent with the assigned structures.


Primary enamines and $\alpha, \beta$-unsaturated carbonyl compounds are the components of the Hantzsch pyridine synthesis [6-8]. Several investigators have reported that primary enaminonitriles react with $\alpha, \beta$-unsaturated carbonyl compounds in the presence of a base to give the pyridines [7, 912]. The Hantzsch synthesis is believed to involve the adduct initially formed by a Michael addition of an enamine to an $\alpha, \beta$-unsaturated carbonyl compound. Generally, the adduct has not been isolated under the reaction conditions
because of its fast intramolecular cyclization to yield the pyridine derivative. Hence, we investigated the conversion of the Michael adducts 2-4 into the pyridines.

Treatment of 2a-d or 3a-c with sodium ethoxide (1 equivalent) in ethanol at room temperature underwent cyclization to give the corresponding 7a-ethoxyhexa-hydrofuro[2,3-b]pyridines 5a-d or 7a-c. Similarly, the reaction of 2a-d or 3a-c with sodium methoxide afforded the 7 a -methoxyhexahydrofuro[2,3-b]pyridines $\mathbf{6 a - d}$ or 8a-c. In contrast, when $\mathbf{3 d}$ was treated with sodium ethoxide or methoxide, a retro-Michael reaction took place to furnish $\mathbf{1 d}$. The ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectra of $\mathbf{5 c}, \mathbf{6 b}, \mathbf{6 c}$ and $\mathbf{8 b}$ show two triplets at near $\delta 1.2$ for a methyl group of an ethoxy group of $\mathbf{5 c}$, two doublets at near $\delta 1.4$ for a 2-methyl group of $\mathbf{6 b}$ and $\mathbf{8 b}$, two singlets at near $\delta 2.1$ for a
cyclization occurred and 4a-d were recovered unchanged. However, under refluxing ethanol, the reaction of 4a-c with sodium ethoxide led to the formation of the dihydrofuro $2,3-b$ ]pyridines 11a-c. In this case, ethanol adducts such as 5 and $\mathbf{6}$ could not be obtained and intramolecular cyclization of $\mathbf{4 d}$ did not proceed.

The formation of 5-8 and $\mathbf{9 - 1 1}$ can be explained in terms of Scheme 2. A Michael addition of $\mathbf{1}$ to $\alpha, \beta$-unsaturated carbonyl compounds gives the adducts 2-4 which undergo intramolecular cyclization and dehydration to form the intermediate dihydropyridines $\mathbf{A}$. Elimination of hydrogen cyanide from $\mathbf{A}$ under basic conditions produces 9-11. Conjugate addition of ethanol or methanol to $\mathbf{A}$ results in the formation of $\mathbf{5}$ and $\mathbf{7}$ or $\mathbf{6}$ and $\mathbf{8}$. Unfortunately, the key intermediate A could not be isolated.


Scheme 2


A


5-8


6-methyl group of $\mathbf{5 c}, \mathbf{6} \mathbf{b}, \mathbf{6 c}$ and $\mathbf{8 b}$, and two singlets at near $\delta 3.6$ for a methyl group of a methoxy group of $\mathbf{6 b}, \mathbf{6 c}$ and $\mathbf{8 b}$. These observations indicate that $5 \mathbf{c}, \mathbf{6 b}, \mathbf{6 c}$ and $\mathbf{8 b}$ exist as two diastereomers, which would probably be formed by an effect of the configuration of methyl or phenyl group at the 2-position, ethoxy or methoxy group and cyano group. The reaction of $\mathbf{1 c}$ with benzalacetone and sodium ethoxide ( 1 equivalent) in ethanol gave $7 \mathbf{c}$ in $42 \%$ yield. In a similar way, by the reaction of $1 \mathbf{c}$ with benzaldehyde and acetone, 7c was also obtained in a somewhat lower yield of $25 \%$. The adducts 5a-d, 6a-d, 7a-c and 8a-c were treated with potassium tert-butoxide in refluxing tert-butyl alcohol to produce the dihydrofuro-[2,3-b]pyridines 9a-d and 10a-c. On the other hand, when a solution of 4a-d with sodium ethoxide (1 equivalent) in ethanol was stirred overnight at room temperature, no

Compound 5d was converted into 6d by treatment with sodium methoxide in refluxing methanol, and conversely, 6d was transformed into 5d by treatment with sodium ethoxide in refluxing ethanol, and 9d was not obtained. This interconversion probably proceeds through the intermediate B (Scheme 2), which would be resistant to elimination of hydrogen cyanide to give 9d due to the absence of phenyl group at the 4-position. Our results show that sodium ethoxide/ethanol or sodium methoxide/methanol system (milder condition) provides a unique possibility to afford Michael adducts 2-4 and alcohol adducts 5-8. Aromatization of 5-8 with potassium tert-butoxide in refluxing tert-butyl alcohol proceeds smoothly to give $\mathbf{9}$ and $\mathbf{1 0}$ in good yields. While, in the case of intramolecular cyclization of 4 , sodium ethoxide in refluxing ethanol system is better than potassium tertbutoxide system (stronger condition).

Subsequently, we undertook a one-pot synthesis of 9-11 from 1 and $\alpha, \beta$-unsaturated carbonyl compounds. Attempts to prepare 9c or 10c from 1c and methyl vinyl ketone or benzalacetone according to the method of Robinson et al. [10] or Shibata et al. [12] were unsuccessful; however, a mixture of 1, benzalacetone or cinnamaldehyde and sodium ethoxide ( 2 equivalents) in ethanol stirred overnight at room temperature, and then refluxed for 1 hour provided 10a-c and 11a-c. In the case of $9 \mathbf{a - d}$, we carried out elimination of hydrogen cyanide by use of potassium tert-butoxide.

## EXPERIMENTAL

All melting points are uncorrected. The ir spectra were recorded on a JASCO FT/IR-230 spectrometer. The ${ }^{1} \mathrm{H} \mathrm{nmr}$ spectra were recorded on a HITACHI R-90 H spectrometer ( 90 MHz ). Chemical shifts ( $\delta$ ) are reported in parts per million ( ppm ) relative to tetramethylsilane as internal standard. Positive FAB mass spectra were obtained on a JEOL JMS-HX 110 spectrometer. Elemental analyses were performed on a HERAUS CHNORAPID analyzer or YANACO MT-6 CHN analyzer.
General Procedure for the Preparation of Michael adducts 2, 3 and 4.

To a solution of sodium ( $23 \mathrm{mg}, 1 \mathrm{mmole}$ ) in anhydrous ethanol ( 2 ml ) were added $\mathbf{1 a - d}$ ( 10 mmoles) and methyl vinyl ketone ( $0.84 \mathrm{~g}, 12$ mmoles), benzalacetone ( $1.75 \mathrm{~g}, 12$ mmoles) or cinnamaldehyde ( $1.58 \mathrm{~g}, 12$ mmoles) and tetrahydrofuran ( 20 ml , in the case of the preparation of $\mathbf{2 a}, \mathbf{b}$ ) or diethyl ether ( 20 ml , 2c, 2d, 3a-d and 4a-d) with stirring. The mixture was stirred at room temperature for 5 hours ( $\mathbf{2 a}, \mathbf{b}$ ) or 2 hours ( $\mathbf{2 c}, \mathbf{2 d}, \mathbf{3 a - d}$ and $\mathbf{4 a - d}$ ). The reaction mixture was neutralized with acetic acid ( 60 $\mathrm{mg}, 1 \mathrm{mmole}$ ) with stirring and ice-cooling, and then cold water was added to the resulting mixture. Further processing of the resulting mixture is described in the following paragraphs.
(A) The resulting mixture was extracted with ethyl acetate. The extract was dried over anhydrous sodium sulfate and concentrated in vacuo. The residue was recrystallized from an appropriate solvent to give 2a, 2b, 2d and $\mathbf{3 b}$.
(B) The precipitate was isolated by filtration, washed with water, dried and recrystallized from an appropriate solvent to yield $\mathbf{2 c}, \mathbf{3 a}, \mathbf{3 c}, 3 d$ and $\mathbf{4 a - d}$.

Tetrahydro-2-imino-3-(3-oxobutyl)-3-furancarbonitrile (2a).
This compound was obtained as colorless prisms ( 1.10 g , $61 \%$ ), mp 75-77 (acetone-petroleum ether); ir (potassium bromide): v $3210(\mathrm{NH}), 2240(\mathrm{C} \equiv \mathrm{N}), 1705(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.09-3.12\left(\mathrm{~m}, 6 \mathrm{H}, 4-\mathrm{H}, 1^{\prime}-\mathrm{H}, 2^{\prime}-\mathrm{H}\right), 2.21$ ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{COCH}_{3}$ ), $4.32(\mathrm{t}, \mathrm{J}=6.5 \mathrm{~Hz}, 2 \mathrm{H}, 5-\mathrm{H}), 7.30 \mathrm{ppm}(\mathrm{br} \mathrm{s}$, $1 \mathrm{H}, \mathrm{NH})$; ms: m/z 181 [M+H] .
Anal. Calcd. for $\mathrm{C}_{19} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, $59.99 ; \mathrm{H}, 6.71 ; \mathrm{N}, 15.35$. Found: C, 60.05; H, 6.43; N, 15.10.

Tetrahydro-2-imino-5-methyl-3-(3-oxobutyl)-3-furancarbonitrile (2b).
This compound was obtained as colorless needles $(0.91 \mathrm{~g}$, $47 \%$ ), mp 102-104 (acetone-petroleum ether); ir (potassium bromide): v $3245(\mathrm{NH}), 2230(\mathrm{C} \equiv \mathrm{N}), 1700(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$
(deuteriochloroform): $\delta 1.41$ (d, J = 6 Hz, 3H, 5-CH3), 1.80 (dd, $\left.\mathrm{J}=10.5,13 \mathrm{~Hz}, 2 \mathrm{H}, 1^{\prime}-\mathrm{H}\right), 2.10-2.37(\mathrm{~m}, 2 \mathrm{H}, 4-\mathrm{H}), 2.21(\mathrm{~s}, 3 \mathrm{H}$, $\mathrm{COCH}_{3}$ ), 2.47-2.81 (m, 2H, 2'-H), 4.43-4.85 (m, 1H, 5-H), 7.24 ppm (br s, 1H, NH); ms: m/z $195[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 61.84; $\mathrm{H}, 7.27$; N, 14.42. Found: C, 61.95; H, 7.11; N, 14.53.

Tetrahydro-2-imino-3-(3-oxobutyl)-5-phenyl-3-furancarbonitrile (2c).

This compound was obtained as colorless needles ( 1.72 g , $67 \%$ ), $\mathrm{mp} 114-116^{\circ}$ (acetone-petroleum ether); ir (potassium bromide): v $3220(\mathrm{NH}), 2230(\mathrm{C} \equiv \mathrm{N}), 1705(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.05-2.62\left(\mathrm{~m}, 4 \mathrm{H}, 4-\mathrm{H}, 1^{\prime}-\mathrm{H}\right), 2.20$ (s, $3 \mathrm{H}, \mathrm{COCH}_{3}$ ), 2.73-3.13 (m, 2H, 2'-H), 5.52 (dd, J $=5.5,10.5$ $\mathrm{Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 7.30-7.36(\mathrm{~m}, 5 \mathrm{H}$, aromatic H$), 7.46 \mathrm{ppm}(\mathrm{br} \mathrm{s}, 1 \mathrm{H}$, NH); ms: m/z 257 [M+H] .

Anal. Calcd. for $\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 70.29; H, 6.29; $\mathrm{N}, 10.93$. Found: C, 69.98; H, 6.37; N, 10.66.
Tetrahydro-2-imino-3-(3-oxobutyl)-4-phenyl-3-furancarbonitrile (2d).

This compound was obtained as colorless needles ( 2.19 g , $86 \%$ ), mp 147-148 (chloroform-petroleum ether); ir (potassium bromide): v $3192(\mathrm{NH}), 2232(\mathrm{C} \equiv \mathrm{N}), 1692(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.09-2.35\left(\mathrm{~m}, 2 \mathrm{H}, 1^{\prime}-\mathrm{H}\right), 2.16(\mathrm{~s}, 3 \mathrm{H}$, $\mathrm{COCH}_{3}$ ), 2.41-3.31 (m, 2H, 2'-H), 3.50 (t, J = $7 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}$ ), 4.49-4.70 (m, 2H, 5-H), 7.29-7.46 ppm (m, 6H, NH, aromatic H); ms : m/z $257[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, $70.29 ; \mathrm{H}, 6.29 ; \mathrm{N}, 10.93$. Found: C, 70.25; H, 6.48; N, 10.70.

Tetrahydro-2-imino-3-(3-oxo-1-phenylbutyl)-3-furancarbonitrile (3a).

This compound was obtained as colorless needles ( $1.24 \mathrm{~g}, 48 \%$ ), $\mathrm{mp} 157-158^{\circ}$ (acetone-petroleum ether); ir (potassium bromide): $v$ $3214(\mathrm{NH})$, $2232(\mathrm{C} \equiv \mathrm{N}), 1707(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 1.65\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{COCH}_{3}\right), 2.06-2.52\left(\mathrm{~m}, 4 \mathrm{H}, 4-\mathrm{H}, 2^{\prime}-\mathrm{H}\right)$, $3.16(\mathrm{t}, \mathrm{J}=8 \mathrm{~Hz}, 1 \mathrm{H}, 1 '-\mathrm{H}), 4.17-4.47(\mathrm{~m}, 2 \mathrm{H}, 5-\mathrm{H}), 7.37(\mathrm{~s}, 5 \mathrm{H}$, aromatic H), $7.37 \mathrm{ppm}(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH}) ; \mathrm{ms}: \mathrm{m} / \mathrm{z} 257[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, $70.29 ; \mathrm{H}, 6.29 ; \mathrm{N}, 10.93$. Found: C, 70.26; H, 6.47; N, 10.73.

Tetrahydro-2-imino-5-methyl-3-(3-oxo-1-phenylbutyl)-3-furancarbonitrile (3b).

This compound was obtained as colorless needles $(1.31 \mathrm{~g}$, $49 \%$ ), mp 162-164 (acetone-petroleum ether); ir (potassium bromide): v $3246(\mathrm{NH}), 2232(\mathrm{C} \equiv \mathrm{N}), 1702(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 1.46$ (d, J $=6 \mathrm{~Hz}, 3 \mathrm{H}, 5-\mathrm{CH}_{3}$ ), 1.63 (s, $\left.3 \mathrm{H}, \mathrm{COCH}_{3}\right), 1.84-2.06(\mathrm{~m}, 1 \mathrm{H}, 4-\mathrm{H}), 2.26-2.58(\mathrm{~m}, 3 \mathrm{H}, 4-\mathrm{H}$, 2'-H), 3.22 (dd, J = 7.5, $\left.16 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{1}^{\prime}-\mathrm{H}\right), 4.55-4.82(\mathrm{~m}, 1 \mathrm{H}, 5-\mathrm{H})$, $7.35(\mathrm{~s}, 5 \mathrm{H}$, aromatic H), $7.35 \mathrm{ppm}(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH})$; ms: m/z 271 $[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 71.09; H, 6.71; N, 10.36. Found: C, 71.11; H, 6.73; N, 10.32.

Tetrahydro-2-imino-3-(3-oxo-1-phenylbutyl)-5-phenyl-3-furancarbonitrile (3c).

This compound was obtained as colorless needles ( 2.75 g , $83 \%$ ), mp 184-185 ${ }^{\circ}$ (chloroform-petroleum ether); ir (potassium bromide): v $3230(\mathrm{NH}), 2240(\mathrm{C} \equiv \mathrm{N}), 1705(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (DMSO-d ${ }_{6}$ ): $\delta 1.49\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{COCH}_{3}\right), 2.09-2.24(\mathrm{~m}, 2 \mathrm{H}, 4-\mathrm{H})$,
2.39-2.89 (m, 2H, 2'-H), $3.51\left(\mathrm{dd}, \mathrm{J}=4.5,11 \mathrm{~Hz}, 1 \mathrm{H}, 1^{\prime}-\mathrm{H}\right), 5.49$ (dd, J = 5, $10.5 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}$ ), 5.61 (br s, 1H, NH), 7.31-7.53 ppm (m, 10 H , aromatic H ); ms: $\mathrm{m} / \mathrm{z} 333[\mathrm{M}+\mathrm{H}]^{+}$.
Anal. Calcd. for $\mathrm{C}_{21} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 75.88; H, 6.06; N, 8.43. Found: C, 75.70; H, 6.00; N, 8.49.
Tetrahydro-2-imino-3-(3-oxo-1-phenylbutyl)-4-phenyl-3-furancarbonitrile (3d).
This compound was obtained as colorless needles $(2.29 \mathrm{~g}$, $69 \%$ ), mp 185-187 ${ }^{\circ}$ (chloroform-petroleum ether); ir (potassium bromide): v $3221(\mathrm{NH}), 2240(\mathrm{C} \equiv \mathrm{N}), 1694(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (DMSO- $\mathrm{d}_{6}$ ): $\delta 1.45\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{COCH}_{3}\right), 1.81-2.37\left(\mathrm{~m}, 2 \mathrm{H}, 2^{\prime}-\mathrm{H}\right)$, $3.41\left(\mathrm{dd}, \mathrm{J}=3.5,12 \mathrm{~Hz}, 1 \mathrm{H}, 1^{\prime}-\mathrm{H}\right), 3.92-4.63(\mathrm{~m}, 3 \mathrm{H}, 4-\mathrm{H}, 5-\mathrm{H})$, 5.66 (br s, 1H, NH), 6.98-7.31 ppm (m, 10H, aromatic H); ms: m/z $333[\mathrm{M}+\mathrm{H}]^{+}$.
Anal. Calcd. for $\mathrm{C}_{21} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, $75.88 ; \mathrm{H}, 6.06 ; \mathrm{N}, 8.43$. Found: C, $75.70 ; \mathrm{H}, 6.18$; N, 8.57.

Tetrahydro-2-imino-3-(3-phenylpropanal-3-yl)-3-furancarbonitrile (4a).

This compound was obtained as colorless prisms ( 1.96 g , $81 \%$ ), mp 204-205 ${ }^{\circ}$ dec (methanol-petroleum ether); ir (potassium bromide): v $3220(\mathrm{NH}), 2230(\mathrm{C} \equiv \mathrm{N}), 1695(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ nmr (DMSO-d ${ }_{6}$ ): $\delta 1.80-2.78\left(\mathrm{~m}, 4 \mathrm{H}, 4-\mathrm{H}, 2^{\prime}-\mathrm{H}\right), 3.29(\mathrm{dd}, \mathrm{J}=3$, $12 \mathrm{~Hz}, 1 \mathrm{H}, 3$ '-H), 3.99-4.53 (m, 2H, 5-H), 5.26-5.36 (m, 1H, CHO ), 5.87 (d, J $=4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{NH}$ ), $7.40 \mathrm{ppm}(\mathrm{s}, 5 \mathrm{H}$, aromatic H ); ms : m/z 243 [M+H] ${ }^{+}$.

Anal. Calcd. for $\mathrm{C}_{14} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, $69.41 ; \mathrm{H}, 5.82 ; \mathrm{N}, 11.56$. Found: C, 69.41; H, 5.89; N, 11.45.

Tetrahydro-2-imino-5-methyl-3-(3-phenylpropanal-3-yl)-3furancarbonitrile (4b).
This compound was obtained as colorless prisms ( 1.25 g , $49 \%$ ), mp 186-187 ${ }^{\circ}$ (acetone-petroleum ether); in (potassium bromide): v $3180(\mathrm{NH}), 2240(\mathrm{C} \equiv \mathrm{N}), 1695(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (DMSO-d ${ }_{6}$ ): $\delta 1.37\left(\mathrm{~d}, \mathrm{~J}=6 \mathrm{~Hz}, 3 \mathrm{H}, 5-\mathrm{CH}_{3}\right), 1.88-2.38(\mathrm{~m}, 4 \mathrm{H}$, $\left.4-\mathrm{H}, 2^{\prime}-\mathrm{H}\right), 3.26$ (dd, J = 3.5, $\left.12.5 \mathrm{~Hz}, 1 \mathrm{H}, 3^{\prime}-\mathrm{H}\right), ~ 4.35-4.79$ (m, $1 \mathrm{H}, 5-\mathrm{H}), 5.03-5.24(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CHO}), 5.99(\mathrm{~d}, \mathrm{~J}=5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{NH})$, $7.37 \mathrm{ppm}(\mathrm{s}, 5 \mathrm{H}$, aromatic H$)$; ms: m/z $257[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{15} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, $70.29 ; \mathrm{H}, 6.29 ; \mathrm{N}, 10.93$. Found: C, 70.29; H, 6.37; N, 10.88 .

Tetrahydro-2-imino-5-phenyl-3-(3-phenylpropanal-3-yl)-3furancarbonitrile (4c).
This compound was obtained as colorless needles $(2.36 \mathrm{~g}$, $74 \%$ ), mp 196-197 ${ }^{\circ}$ (chloroform-petroleum ether); ir (potassium bromide): v $3210(\mathrm{NH}), 2240(\mathrm{C} \equiv \mathrm{N}), 1690(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (DMSO-d ${ }_{6}$ ): $\delta 1.89-2.73\left(\mathrm{~m}, 4 \mathrm{H}, 4-\mathrm{H}, 2^{\prime}-\mathrm{H}\right), 3.45(\mathrm{dd}, \mathrm{J}=3.5$, $\left.12.5 \mathrm{~Hz}, 1 \mathrm{H}, 3^{\prime}-\mathrm{H}\right), 5.13-5.29$ (m, 1H, CHO), 5.42-5.82 (m, 1H, $5-\mathrm{H}), 6.08$ (br s, $1 \mathrm{H}, \mathrm{NH}$ ), $7.23-7.53 \mathrm{ppm}(\mathrm{m}, 10 \mathrm{H}$, aromatic H); ms : m/z $319[\mathrm{M}+\mathrm{H}]^{+}$.
Anal. Calcd. for $\mathrm{C}_{20} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{2} \cdot 0.1 \mathrm{H}_{2} \mathrm{O}$ : C, 75.03 ; $\mathrm{H}, 5.73$; N , 8.75. Found: C, $74.95 ; \mathrm{H}, 5.75 ; \mathrm{N}, 8.73$.

Tetrahydro-2-imino-4-phenyl-3-(3-phenylpropanal-3-yl)-3-furancarbonitrile (4d).

This compound was obtained as colorless needles $(2.45 \mathrm{~g}$, $77 \%$ ), mp 200-201 ${ }^{\circ}$ (chloroform-petroleum ether); ir (potassium bromide): v $3165(\mathrm{NH}), 2230(\mathrm{C} \equiv \mathrm{N}), 1690(\mathrm{C}=\mathrm{O}) \mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}\left(\mathrm{DMSO}_{-} \mathrm{d}_{6}\right): \delta 1.72-1.92\left(\mathrm{~m}, 1 \mathrm{H}, 2^{\prime}-\mathrm{H}\right), 2.45(\mathrm{dd}, \mathrm{J}=$ $5,13 \mathrm{~Hz}, 1 \mathrm{H}, 2^{\prime}-\mathrm{H}$ ), 3.44 (dd, J = 3.5, $12.5 \mathrm{~Hz}, 1 \mathrm{H}, 3^{\prime}-\mathrm{H}$ ),
3.95-4.64 (m, 3H, 4-H, 5-H), 5.20-5.35 (m, 1H, CHO), 5.96 (d, $\mathrm{J}=3 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{NH}), 6.97-7.32 \mathrm{ppm}(\mathrm{m}, 10 \mathrm{H}$, aromatic H$)$; ms: m/z $319[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{20} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 75.45 ; $\mathrm{H}, 5.70 ; \mathrm{N}, 8.80$. Found: C, 75.45; H, 5.65; N, 8.86.
General Procedure for the Preparation of 5a-d, 6a-d, 7a-c and 8a-c from 2a-d or 3a-c.

A mixture of 2a-d or 3a-c ( 10 mmoles ) in a solution of sodium ( $0.23 \mathrm{~g}, 10 \mathrm{mmoles}$ ) in anhydrous ethanol ( 20 ml ) or anhydrous methanol $(20 \mathrm{ml})$ was stirred at room temperature overnight, and then cold water was added to the reaction mixture. The resulting mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and concentrated in vacuo. The residue was purified by column chromatography on silica gel with methylene chloride as the eluent to give 5a-d, 6a-d, 7a-c and 8a-c.

7a-Ethoxy-2,3,3a,4,5,7a-hexahydro-6-methylfuro[2,3-b]pyri-dine-3a-carbonitrile (5a).

This compound was obtained as colorless plates $(0.79 \mathrm{~g}, 38 \%)$, $\mathrm{mp} 46-48^{\circ}$ (diethyl ether-petroleum ether); ir (potassium bromide): v $2240(\mathrm{C} \equiv \mathrm{N}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 1.23$ (t, J $=7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{3}$ ), 1.63-2.95 (m, 6H, 3-H, 4-H, 5-H), $2.06\left(\mathrm{~s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}\right), 3.78-4.12 \mathrm{ppm}\left(\mathrm{m}, 4 \mathrm{H}, 2-\mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{3}\right)$; ms : m/z $209[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 63.44; $\mathrm{H}, 7.74 ; \mathrm{N}, 13.45$. Found: C, 63.65; H, 7.59; N, 13.23.

7a-Ethoxy-2,3,3a,4,5,7a-hexahydro-2,6-dimethylfuro[2,3-b]-pyridine-3a-carbonitrile ( $\mathbf{5 b}$ ).

This compound was obtained as colorless needles $(0.84 \mathrm{~g}$, $38 \%$ ), mp 54-55 (diethyl ether-petroleum ether); ir (potassium bromide): v $2242(\mathrm{C} \equiv \mathrm{N}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta$ $1.23\left(\mathrm{t}, \mathrm{J}=7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 1.30(\mathrm{~d}, \mathrm{~J}=6.5 \mathrm{~Hz}, 3 \mathrm{H}$, $2-\mathrm{CH}_{3}$ ), 1.56-2.48 (m, 5H, 3-H, 4-H, 5-H), 2.07 ( $\mathrm{s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}$ ), $2.88(\mathrm{dd}, \mathrm{J}=8.5,12.5 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 3.76-4.48 \mathrm{ppm}(\mathrm{m}, 3 \mathrm{H}, 2-\mathrm{H}$, $\mathrm{OCH}_{2} \mathrm{CH}_{3}$ ); ms: m/z $223[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{12} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{2}: \mathrm{C}, 64.84 ; \mathrm{H}, 8.16 ; \mathrm{N}, 12.60$. Found: C, 64.64; H, 8.16; N, 12.38.
7a-Ethoxy-2,3,3a, 4,5,7a-hexahydro-6-methyl-2-phenylfuro-[2,3-b]pyridine-3a-carbonitrile (5c).

This compound was obtained as colorless needles $(1.50 \mathrm{~g}$, $53 \%$ ), mp 102-104 (acetone-petroleum ether); ir (potassium bromide): $v 2241(\mathrm{C} \equiv \mathrm{N}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 1.17$ and $1.28\left(\mathrm{t}, \mathrm{J}=7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 1.69-2.18(\mathrm{~m}, 3 \mathrm{H}, 3-\mathrm{H}$, $4-\mathrm{H}), 2.09$ and $2.11\left(\mathrm{~s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}\right), 2.30-2.64(\mathrm{~m}, 2 \mathrm{H}, 3-\mathrm{H}, 5-\mathrm{H})$, 2.97 and $3.19(\mathrm{dd}, \mathrm{J}=10,12 \mathrm{~Hz}, \mathrm{dd}, \mathrm{J}=8.5,12.5 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H})$, 3.78-4.25 (m, $2 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{3}$ ), 5.09-5.32 (m, 1H, 2-H), 7.31-7.36 ppm (m, 5 H , aromatic H ); ms: m/z $285[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{17} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 71.81; H, 7.09; N, 9.85. Found: C, 72.07; H, 7.08; N, 9.94.

7a-Ethoxy-2,3,3a, 4,5,7a-hexahydro-6-methyl-3-phenylfuro-[2,3-b]pyridine-3a-carbonitrile (5d).

This compound was obtained as colorless needles $(1.36 \mathrm{~g}$, $48 \%$ ) $\mathrm{mp} 138-140^{\circ}$ (acetone-petroleum ether); ir (potassium bromide): v $2242(\mathrm{C} \equiv \mathrm{N}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta$ $1.34\left(\mathrm{t}, \mathrm{J}=7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 1.70-2.09(\mathrm{~m}, 2 \mathrm{H}, 4-\mathrm{H}), 2.09$ ( $\mathrm{s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}$ ), 2.31-2.50 (m, 2H, 5-H), $3.34(\mathrm{t}, \mathrm{J}=8.5 \mathrm{~Hz}, 1 \mathrm{H}$,
$3-\mathrm{H}), 3.94-4.44\left(\mathrm{~m}, 4 \mathrm{H}, 2-\mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 7.31-7.49 \mathrm{ppm}(\mathrm{m}, 5 \mathrm{H}$, aromatic H$)$; ms: m/z $285[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{17} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 71.81; H, 7.09; N, 9.85. Found: C, 72.07; H, 7.05; N, 9.79.

2,3,3a,4,5,7a-Hexahydro-7a-methoxy-6-methylfuro[2,3-b]pyri-dine-3a-carbonitrile (6a).

This compound was obtained as colorless columns $(0.23 \mathrm{~g}$, $12 \%$ ), mp 62-63 ${ }^{\circ}$ (diethyl ether-petroleum ether); ir (potassium bromide): v $2240(\mathrm{C} \equiv \mathrm{N}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta$ $1.67-2.94(\mathrm{~m}, 6 \mathrm{H}, 3-\mathrm{H}, 4-\mathrm{H}, 5-\mathrm{H}), 2.08\left(\mathrm{~s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}\right), 3.56(\mathrm{~s}$, $\left.3 \mathrm{H}, \mathrm{OCH}_{3}\right), 3.99 \mathrm{ppm}(\mathrm{dd}, \mathrm{J}=6.5,8.5 \mathrm{~Hz}, 2 \mathrm{H}, 2-\mathrm{H}) ; \mathrm{ms}: \mathrm{m} / \mathrm{z}$ $195[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{10} \mathrm{H}_{14} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 61.84; H, 7.27; N, 14.42. Found: C, 62.13; H, 7.18; N, 14.25.

2,3,3a,4,5,7a-Hexahydro-7a-methoxy-2,6-dimethylfuro[2,3-b]-pyridine-3a-carbonitrile ( $\mathbf{6 b}$ ).

This compound was obtained as colorless prisms $(0.25 \mathrm{~g}$, $12 \%$ ), mp 68-69 (diethyl ether-petroleum ether); ir (potassium bromide): v $2242(\mathrm{C} \equiv \mathrm{N}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta$ 1.31 and $1.37\left(\mathrm{t}, \mathrm{J}=6.5 \mathrm{~Hz}, 3 \mathrm{H}, 2-\mathrm{CH}_{3}\right), 1.58-1.98(\mathrm{~m}, 2 \mathrm{H}, 4-\mathrm{H})$, 2.06 and $2.08\left(\mathrm{~s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}\right), 2.28-2.49(\mathrm{~m}, 3 \mathrm{H}, 3-\mathrm{H}, 5-\mathrm{H}), 2.87$ $(\mathrm{dd}, \mathrm{J}=8.5,12.5 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 3.54$ and $3.58\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OCH}_{3}\right)$, 4.15-4.46 ppm (m, 1H, 2-H); ms: m/z $209[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 63.44; H, 7.74; N, 13.45. Found: C, 63.70; H, 7.96; N, 13.77.

2,3,3a,4,5,7a-Hexahydro-7a-methoxy-6-methyl-2-phenylfuro $[2,3-b]$ pyridine-3a-carbonitrile ( $\mathbf{6 c}$ ).

This compound was obtained as colorless needles $(0.68 \mathrm{~g}$, $25 \%$ ), $\mathrm{mp} 98-100^{\circ}$ (acetone-petroleum ether); ir (potassium bromide): v $2236(\mathrm{C} \equiv \mathrm{N}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 1.90-$ $2.15(\mathrm{~m}, 2 \mathrm{H}, 4-\mathrm{H}), 2.10$ and $2.13\left(\mathrm{~s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}\right), 2.31-2.65(\mathrm{~m}$, $3 \mathrm{H}, 3-\mathrm{H}, 5-\mathrm{H}), 2.85-3.30(\mathrm{~m}, 1 \mathrm{H}, 5-\mathrm{H}), 3.60$ and $3.63(\mathrm{~s}, 3 \mathrm{H}$, $\left.\mathrm{OCH}_{3}\right), 5.09-5.28(\mathrm{~m}, 1 \mathrm{H}, 2-\mathrm{H}), 7.26-7.37 \mathrm{ppm}(\mathrm{m}, 5 \mathrm{H}$, aromatic H); ms: m/z $271[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 71.09; $\mathrm{H}, 6.71 ; \mathrm{N}, 10.36$. Found: C, 71.36; H, 6.43; N, 10.54.

2,3,3a,4,5,7a-Hexahydro-7a-methoxy-6-methyl-3-phenyl-furo[2,3-b]pyridine-3a-carbonitrile ( $\mathbf{6 d}$ ).

This compound was obtained as colorless needles $(0.90 \mathrm{~g}$, $33 \%$ ), $\mathrm{mp} 167-169^{\circ}$ (acetone-petroleum ether); ir (potassium bromide): v $2242(\mathrm{C} \equiv \mathrm{N}) \mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 1.84-$ $2.17(\mathrm{~m}, 2 \mathrm{H}, 4-\mathrm{H}), 2.11\left(\mathrm{~s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}\right), 2.32-2.52(\mathrm{~m}, 2 \mathrm{H}, 5-\mathrm{H})$, $3.36(\mathrm{t}, \mathrm{J}=9 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}), 3.66\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OCH}_{3}\right), 4.01(\mathrm{t}, \mathrm{J}=9$ $\mathrm{Hz}, 1 \mathrm{H}, 2-\mathrm{H}), 4.35(\mathrm{t}, \mathrm{J}=9 \mathrm{~Hz}, 1 \mathrm{H}, 2-\mathrm{H}), 7.25-7.47 \mathrm{ppm}(\mathrm{m}, 5 \mathrm{H}$, aromatic H$)$; ms: m/z $271[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, $71.09 ; \mathrm{H}, 6.71 ; \mathrm{N}, 10.36$. Found: C, 71.10; H, 6.71; N, 10.39.

7a-Ethoxy-2,3,3a,4,5,7a-hexahydro-6-methyl-4-phenylfuro-[2,3-b]pyridine-3a-carbonitrile (7a).

This compound was obtained as colorless prisms $(0.54 \mathrm{~g}$, $19 \%$ ), mp 114-116 ${ }^{\circ}$ (acetone-petroleum ether); ir (potassium bromide): $\vee 2240(\mathrm{C} \equiv \mathrm{N}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 1.25$ $\left(\mathrm{t}, \mathrm{J}=7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 2.12\left(\mathrm{~s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}\right), 2.20-2.94(\mathrm{~m}$, $5 \mathrm{H}, 3-\mathrm{H}, 4-\mathrm{H}, 5-\mathrm{H}), 3.93-4.13\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 7.37 \mathrm{ppm}(\mathrm{s}$, 5 H , aromatic H$) ; \mathrm{ms}: \mathrm{m} / \mathrm{z} 285[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{17} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, $71.81 ; \mathrm{H}, 7.09 ; \mathrm{N}, 9.85$. Found: C, 71.81; H, 7.09; N, 9.87.
7a-Ethoxy-2,3,3a,4,5,7a-hexahydro-2,6-dimethyl-4-phenyl-furo[2,3-b]pyridine-3a-carbonitrile (7b).

This compound was obtained as colorless columns $(0.53 \mathrm{~g}$, $18 \%$ ), mp 122-124 ${ }^{\circ}$ (acetone-petroleum ether); ir (potassium bromide): $\vee 2232(\mathrm{C} \equiv \mathrm{N}) \mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 1.25$ $\left(\mathrm{t}, \mathrm{J}=7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 1.37\left(\mathrm{~d}, \mathrm{~J}=6 \mathrm{~Hz}, 3 \mathrm{H}, 2-\mathrm{CH}_{3}\right), 2.10$ (s, 3H, 6-CH3), 2.22-2.99 (m, 5H, 3-H, 4-H, 5-H), 3.86-4.24 (m, $\left.2 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 4.34-4.59(\mathrm{~m}, 1 \mathrm{H}, 2-\mathrm{H}), 7.39 \mathrm{ppm}(\mathrm{s}, 5 \mathrm{H}$, aromatic H); ms: m/z $299[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{18} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 72.46; H, 7.43; N, 9.39. Found: C, 72.47; H, 7.49; N, 9.38.

7a-Ethoxy-2,3,3a,4,5,7a-hexahydro-6-methyl-2,4-diphenyl-furo[2,3-b]pyridine-3a-carbonitrile (7c).

This compound was obtained as colorless prisms $(1.55 \mathrm{~g}$, $43 \%$ ), mp 188-190 ${ }^{\circ}$ (acetone-petroleum ether); ir (potassium bromide): v $2251(\mathrm{C} \equiv \mathrm{N}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta$ $1.17\left(\mathrm{t}, \mathrm{J}=7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 2.15\left(\mathrm{~s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}\right), 2.37-2.89$ $(\mathrm{m}, 4 \mathrm{H}, 3-\mathrm{H}, 4-\mathrm{H}, 5-\mathrm{H}), 3.07-3.31(\mathrm{~m}, 1 \mathrm{H}, 4-\mathrm{H}), 3.71-4.35(\mathrm{~m}$, $\left.2 \mathrm{H}, \mathrm{OCH}_{2} \mathrm{CH}_{3}\right), 5.25-5.44(\mathrm{~m}, 1 \mathrm{H}, 2-\mathrm{H}), 7.25-7.50 \mathrm{ppm}(\mathrm{m}$, 10 H , aromatic H$) ; \mathrm{ms}: \mathrm{m} / \mathrm{z} 361[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{23} \mathrm{H}_{24} \mathrm{~N}_{2} \mathrm{O}_{2}: \mathrm{C}, 76.64 ; \mathrm{H}, 6.71 ; \mathrm{N}, 7.77$. Found: C, 76.92; H, 6.63; N, 7.86.

2,3,3a,4,5,7a-Hexahydro-7a-methoxy-6-methyl-4-phenylfuro $2,3-b]$ pyridine-3a-carbonitrile ( $\mathbf{8 a}$ ).

This compound was obtained as colorless prisms $(0.52 \mathrm{~g}$, $19 \%$ ), mp 153-154 ${ }^{\circ}$ (acetone-petroleum ether); ir (potassium bromide): $\vee 2241(\mathrm{C} \equiv \mathrm{N}) \mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.13$ (s, 3H, 6-CH3 ), 2.13-2.95 (m, 5H, $3-\mathrm{H}, 4-\mathrm{H}, 5-\mathrm{H}), 3.61(\mathrm{~s}, 3 \mathrm{H}$, $\left.\mathrm{OCH}_{3}\right), 3.96-4.14(\mathrm{~m}, 2 \mathrm{H}, 2-\mathrm{H}), 7.37 \mathrm{ppm}(\mathrm{s}, 5 \mathrm{H}$, aromatic H$)$; $\mathrm{ms}: \mathrm{m} / \mathrm{z} 271[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{16} \mathrm{H}_{18} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, $71.09 ; \mathrm{H}, 6.71 ; \mathrm{N}, 10.36$. Found: C, 71.12; H, 6.72; N, 10.33.

2,3,3a,4,5,7a-Hexahydro-7a-methoxy-2,6-dimethyl-4-phenyl-furo[2,3-b]pyridine-3a-carbonitrile ( $\mathbf{8 b}$ ).

This compound was obtained as colorless prisms $(0.42 \mathrm{~g}$, $15 \%$ ), mp 151-153 ${ }^{\circ}$ (acetone-petroleum ether); ir (potassium bromide): $\vee 2242(\mathrm{C} \equiv \mathrm{N}) \mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 1.38$ and $1.39\left(\mathrm{~d}, \mathrm{~J}=6.5 \mathrm{~Hz}, 3 \mathrm{H}, 2-\mathrm{CH}_{3}\right), 2.12$ and $2.13(\mathrm{~s}, 3 \mathrm{H}, 6-$ $\left.\mathrm{CH}_{3}\right), 1.72-3.07(\mathrm{~m}, 5 \mathrm{H}, 3-\mathrm{H}, 4-\mathrm{H}, 5-\mathrm{H}), 3.60$ and $3.63(\mathrm{~s}, 3 \mathrm{H}$, $\left.\mathrm{OCH}_{3}\right), 4.18-4.60(\mathrm{~m}, 1 \mathrm{H}, 2-\mathrm{H}), 7.37 \mathrm{ppm}(\mathrm{s}, 5 \mathrm{H}$, aromatic H$)$; ms: m/z $285[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{17} \mathrm{H}_{20} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, $71.81 ; \mathrm{H}, 7.09 ; \mathrm{N}, 9.85$. Found: C, 71.86; H, 7.08; N, 9.87.
2,3,3a,4,5,7a-Hexahydro-7a-methoxy-6-methyl-2,4-diphenylfuro $2,3-b]$ pyridine-3a-carbonitrile (8c).

This compound was obtained as colorless prisms $(0.48 \mathrm{~g}$, $14 \%$ ), mp 188-190 ${ }^{\circ}$ (acetone-petroleum ether); ir (potassium bromide): $\vee 2242(\mathrm{C} \equiv \mathrm{N}) \mathrm{cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.19$ $\left(\mathrm{s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}\right), 2.11-3.09(\mathrm{~m}, 5 \mathrm{H}, 3-\mathrm{H}, 4-\mathrm{H}, 5-\mathrm{H}), 3.70(\mathrm{~s}, 3 \mathrm{H}$, $\left.\mathrm{OCH}_{3}\right), 5.25(\mathrm{dd}, \mathrm{J}=5,8.5 \mathrm{~Hz}, 1 \mathrm{H}, 2-\mathrm{H}), 6.97-7.08(\mathrm{~m}, 2 \mathrm{H}$, aromatic H$), 7.17-7.33(\mathrm{~m}, 3 \mathrm{H}$, aromatic H$), 7.37 \mathrm{ppm}(\mathrm{s}, 5 \mathrm{H}$, aromatic H); ms: m/z $347[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{22} \mathrm{H}_{22} \mathrm{~N}_{2} \mathrm{O}_{2}$ : C, 76.28; H, 6.40; N, 8.09. Found: C, 76.29; H, 6.47; N, 8.05.

The Preparation of $\mathbf{7 c}$ from $\mathbf{1 c}$ and Benzalacetone.
A mixture of $\mathbf{1 c}(1.86 \mathrm{~g}, 10 \mathrm{mmoles})$, benzalacetone ( 1.75 g , 12 mmoles) in a solution of sodium ( $0.23 \mathrm{~g}, 10 \mathrm{mmoles}$ ) in anhydrous ethanol ( 20 ml ) was stirred at room temperature overnight, and then acetic acid $(0.60 \mathrm{~g}, 10 \mathrm{mmoles})$ was added to the reaction mixture. The solvent was removed in vacuo, and then cold water was added to the residue. The resulting mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and concentrated in vacuo. The residue was purified by column chromatography on silica gel with methylene chloride as the eluent to give $7 \mathrm{c}(1.50 \mathrm{~g}, 42 \%)$. The melting point and ir spectrum of this compound coincided with those of an authentic sample prepared from $3 \mathbf{c}$ and sodium ethoxide.
The Preparation of $\mathbf{7 c}$ from $\mathbf{1 c}$, Benzaldehyde and Acetone.
A mixture of $\mathbf{1 c}(1.86 \mathrm{~g}, 10$ mmoles $)$, benzaldehyde $(1.59 \mathrm{~g}, 15$ mmoles), acetone ( $1.59 \mathrm{~g}, 20 \mathrm{mmoles}$ ) in a solution of sodium ( $0.23 \mathrm{~g}, 10 \mathrm{mmoles}$ ) in anhydrous ethanol ( 20 ml ) was stirred at room temperature overnight. After work-up as described above, compound $7 \mathbf{c}(0.89 \mathrm{~g}, 25 \%)$ was obtained. The melting point and ir spectrum of this compound coincided with those of an authentic sample prepared from $\mathbf{3 c}$ and sodium ethoxide.
General Procedure for the Preparation of 9a-d and 10a-c from 5a-d, 6a-d or 7a-c, 8a-c.

A solution of 5a-d, 6a-d, 7a-c or 8a-c (10 mmoles) and potassium tert-butoxide ( $1.12 \mathrm{~g}, 10 \mathrm{mmoles}$ ) in anhydrous tert-butyl alcohol ( 30 ml ) was refluxed for 1 hour. After removal of the solvent in vacuo, cold water was added to the residue. The resulting mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and concentrated in vacuo. The residue was purified by column chromatography on silica gel with methylene chloride as the eluent to yield $9 \mathbf{a - d}$ and 10a-c.

## 2,3-Dihydro-6-methylfuro[2,3-b]pyridine (9a).

This compound was obtained as colorless plates [from 5a: 0.88 $\mathrm{g}(65 \%)$, from 6a: $0.92 \mathrm{~g}(68 \%)$ ], mp 41-42 ${ }^{\circ}$ (diethyl ether-petroleum ether); ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.41\left(\mathrm{~s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}\right)$, 3.19 (t, J = $8.5 \mathrm{~Hz}, 2 \mathrm{H}, 3-\mathrm{H}), 4.59(\mathrm{t}, \mathrm{J}=8.5 \mathrm{~Hz}, 2 \mathrm{H}, 2-\mathrm{H}), 6.62$ (d, J = $7 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}$ ), $7.34 \mathrm{ppm}(\mathrm{d}, \mathrm{J}=7 \mathrm{~Hz}, 1 \mathrm{H}, 4-\mathrm{H}) ; \mathrm{ms}: \mathrm{m} / \mathrm{z}$ $136[\mathrm{M}+\mathrm{H}]^{+}$.
Anal. Calcd. for $\mathrm{C}_{8} \mathrm{H}_{9} \mathrm{NO}: \mathrm{C}, 71.09 ; \mathrm{H}, 6.71 ; \mathrm{N}, 10.36$. Found: C, 71.02; H, 6.75; N, 10.37.

## 2,3-Dihydro-2,6-dimethylfuro[2,3-b]pyridine (9b).

This compound was obtained as pale yellow oil [from 5b: 1.08 g ( $72 \%$ ), from 6b: $0.93 \mathrm{~g}(62 \%)$ ]; ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta$ $1.47\left(\mathrm{~d}, \mathrm{~J}=6 \mathrm{~Hz}, 3 \mathrm{H}, 2-\mathrm{CH}_{3}\right), 2.41\left(\mathrm{~s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}\right), 2.66(\mathrm{dd}, \mathrm{J}=$ $7,15.5 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}$ ), 3.31 (dd, J = $8.5,15.5 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}), 4.75-$ $5.13(\mathrm{~m}, 1 \mathrm{H}, 2-\mathrm{H}), 6.60(\mathrm{~d}, \mathrm{~J}=7 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 7.30 \mathrm{ppm}(\mathrm{d}, \mathrm{J}=7$ $\mathrm{Hz}, 1 \mathrm{H}, 4-\mathrm{H}) ; \mathrm{ms}$ : m/z $150[\mathrm{M}+\mathrm{H}]^{+}$.
Anal. Calcd. for $\mathrm{C}_{9} \mathrm{H}_{11} \mathrm{NO}: \mathrm{C}, 72.46 ; \mathrm{H}, 7.43 ; \mathrm{N}, 9.39$. Found: C, 72.19; H, 7.46; N, 9.24.
2,3-Dihydro-6-methyl-2-phenylfuro[2,3-b]pyridine (9c).
This compound was obtained as colorless prisms [from 5c: 1.62 $\mathrm{g}(77 \%)$, from 6c: $1.48 \mathrm{~g}(70 \%)$ ], mp 66-67 ${ }^{\circ}$ (diethyl ether-petroleum ether); ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.46\left(\mathrm{~s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}\right)$, 3.13 (dd, J = 8, $16 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}$ ), 3.64 (dd, $\mathrm{J}=9.5,16 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}$ ), $5.80(\mathrm{dd}, \mathrm{J}=8,9.5 \mathrm{~Hz}, 1 \mathrm{H}, 2-\mathrm{H}), 6.66(\mathrm{~d}, \mathrm{~J}=7 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 7.26-$ $7.43 \mathrm{ppm}\left(\mathrm{m}, 6 \mathrm{H}, 4-\mathrm{H}\right.$, aromatic H ); ms: m/z $212[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{NO}: \mathrm{C}, 79.59 ; \mathrm{H}, 6.20 ; \mathrm{N}, 6.63$. Found: C, 79.52; H, 6.17; N, 6.42.

2,3-Dihydro-6-methyl-3-phenylfuro[2,3-b]pyridine (9d).
This compound was obtained as colorless prisms [from 5d: $1.48 \mathrm{~g}(70 \%)$, from $\mathbf{6 d}: 1.27 \mathrm{~g}(60 \%)$ ], mp 67-68 ${ }^{\circ}$ (diethyl etherpetroleum ether); ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.46$ (s, 3H, 6$\mathrm{CH}_{3}$ ), 4.35-4.73 ( $\mathrm{m}, 2 \mathrm{H}, 2-\mathrm{H}$ ), 4.94 (dd, $\mathrm{J}=7,8.5 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}$ ), $6.66(\mathrm{~d}, \mathrm{~J}=7 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 7.14-7.36 \mathrm{ppm}(\mathrm{m}, 6 \mathrm{H}, 4-\mathrm{H}$, aromatic $\mathrm{H})$; ms: m/z $212[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{NO}: \mathrm{C}, 79.59 ; \mathrm{H}, 6.20 ; \mathrm{N}, 6.63$. Found: C, 79.49; H, 6.30; N, 6.71 .

## 2,3-Dihydro-6-methyl-4-phenylfuro[2,3-b]pyridine (10a).

This compound was obtained as colorless needles [from 7a: $1.89 \mathrm{~g}(90 \%)$, from 8a: $1.71 \mathrm{~g}(80 \%)], \mathrm{mp} 105-106^{\circ}$ (acetonepetroleum ether); ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 2.47$ (s, 3H, 6$\mathrm{CH}_{3}$ ), $3.30(\mathrm{t}, \mathrm{J}=8.5 \mathrm{~Hz}, 2 \mathrm{H}, 3-\mathrm{H}), 4.60(\mathrm{t}, \mathrm{J}=8.5 \mathrm{~Hz}, 2 \mathrm{H}, 2-\mathrm{H})$, $6.75(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 7.45 \mathrm{ppm}(\mathrm{s}, 5 \mathrm{H}$, aromatic H$) ; \mathrm{ms}: \mathrm{m} / \mathrm{z} 212$ $[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{NO}: \mathrm{C}, 79.59 ; \mathrm{H}, 6.20 ; \mathrm{N}, 6.63$. Found: C, 79.69; H, 6.18; N, 6.58 .

2,3-Dihydro-2,6-dimethyl-4-phenylfuro[2,3-b]pyridine (10b).
This compound was obtained as colorless needles [from 7b: $2.08 \mathrm{~g}(92 \%)$, from 8b: $1.83 \mathrm{~g}(81 \%)], \mathrm{mp} 115-117^{\circ}$ (acetonepetroleum ether); ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 1.48$ (d, J = 6 $\left.\mathrm{Hz}, 3 \mathrm{H}, 2-\mathrm{CH}_{3}\right), 2.47\left(\mathrm{~s}, 3 \mathrm{H}, 6-\mathrm{CH}_{3}\right), 2.88(\mathrm{dd}, \mathrm{J}=7,16 \mathrm{~Hz}, 1 \mathrm{H}$, $3-\mathrm{H}), 3.44(\mathrm{dd}, \mathrm{J}=9.5,16 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}), 4.77-5.17(\mathrm{~m}, 1 \mathrm{H}, 2-\mathrm{H})$, $6.74(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H}), 7.44 \mathrm{ppm}(\mathrm{s}, 5 \mathrm{H}$, aromatic H$) ; \mathrm{ms}: \mathrm{m} / \mathrm{z} 226$ $[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{NO}$ : C, 79.97; $\mathrm{H}, 6.71 ; \mathrm{N}, 6.22$. Found: C, 80.07; H, 6.72; N, 6.25 .

## 2,3-Dihydro-6-methyl-2,4-diphenylfuro[2,3-b]pyridine (10c).

This compound was obtained as colorless needles [from $7 \mathbf{c}$ : $2.62 \mathrm{~g}(91 \%)$, from 8c: $2.61 \mathrm{~g}(91 \%)], \mathrm{mp} 119-121^{\circ}$ (acetonepetroleum ether); ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\boldsymbol{\delta} 2.51$ (s, 3H, 6$\mathrm{CH}_{3}$ ), 3.27 (dd, J = 8, $16 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}$ ), 3.76 (dd, J $=9.5,16 \mathrm{~Hz}$, $1 \mathrm{H}, 3-\mathrm{H}), 5.82(\mathrm{dd}, \mathrm{J}=8,9.5 \mathrm{~Hz}, 1 \mathrm{H}, 2-\mathrm{H}), 6.80(\mathrm{~s}, 1 \mathrm{H}, 5-\mathrm{H})$, $7.25-7.51 \mathrm{ppm}(\mathrm{m}, 10 \mathrm{H}$, aromatic H$)$; ms: m/z $288[\mathrm{M}+\mathrm{H}]^{+}$.
Anal. Calcd. for $\mathrm{C}_{20} \mathrm{H}_{17} \mathrm{NO}$ : C, 83.59; H, 5.96; N, 4.87. Found: C, 83.47; H, 5.96; N, 4.90 .
General Procedure for the Preparation of 11a-c from 4a-c and Sodium Ethoxide.

A mixture of 4a-c ( 10 mmoles ) in a solution of sodium $(0.23 \mathrm{~g}$, 10 mmoles) in anhydrous ethanol ( 30 ml ) was refluxed for 2 hours. After removal of the solvent in vacuo, cold water was added to the residue. The resulting mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and concentrated in vacuo. The residue was purified by column chromatography on silica gel with methylene chloride as the eluent to give 11a-c.

## 2,3-Dihydro-4-phenylfuro[2,3-b]pyridine (11a).

This compound was obtained as colorless columns $(1.16 \mathrm{~g}$, $59 \%$ ), mp 77-78 (diethyl ether-petroleum ether); ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 3.36$ (t, J = $8.5 \mathrm{~Hz}, 2 \mathrm{H}, 3-\mathrm{H}), 4.62(\mathrm{t}, \mathrm{J}=8.5$ $\mathrm{Hz}, 2 \mathrm{H}, 2-\mathrm{H}), 6.88(\mathrm{~d}, \mathrm{~J}=5.5 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 7.47(\mathrm{~s}, 5 \mathrm{H}$, aromatic H), $8.05 \mathrm{ppm}(\mathrm{d}, \mathrm{J}=5.5 \mathrm{~Hz}, 1 \mathrm{H}, 6-\mathrm{H})$; ms: m/z $198[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{13} \mathrm{H}_{11} \mathrm{NO}: \mathrm{C}, 79.17 ; \mathrm{H}, 5.62 ; \mathrm{N}, 7.10$. Found: C, 79.16; H, 5.73; N, 7.12.

## 2,3-Dihydro-2-methyl-4-phenylfuro[2,3-b]pyridine (11b).

This compound was obtained as colorless needles $(0.97 \mathrm{~g}$, $46 \%$ ), mp $55-56^{\circ}$ (diethyl ether-petroleum ether); ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 1.50\left(\mathrm{~d}, \mathrm{~J}=6 \mathrm{~Hz}, 3 \mathrm{H}, 2-\mathrm{CH}_{3}\right), 2.94$ (dd, J = $7.5,16.5 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}$ ), 3.48 (dd, J = 8, $16.5 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}$ ), $4.77-$ $5.16(\mathrm{~m}, 1 \mathrm{H}, 2-\mathrm{H}), 6.87(\mathrm{~d}, \mathrm{~J}=5.5 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 7.46(\mathrm{~s}, 5 \mathrm{H}$, aromatic H), $8.04 \mathrm{ppm}(\mathrm{d}, \mathrm{J}=5.5 \mathrm{~Hz}, 1 \mathrm{H}, 6-\mathrm{H}$ ); ms: m/z 212 $[\mathrm{M}+\mathrm{H}]^{+}$.

Anal. Calcd. for $\mathrm{C}_{14} \mathrm{H}_{13} \mathrm{NO}: \mathrm{C}, 79.59 ; \mathrm{H}, 6.20 ; \mathrm{N}, 6.63$. Found: C, 79.59; H, 6.26; N, 6.65.

## 2,3-Dihydro-2,4-diphenylfuro[2,3-b]pyridine (11c).

This compound was obtained as colorless needles $(0.77 \mathrm{~g}$, $28 \%$ ), mp 120-122 ${ }^{\circ}$ (acetone-petroleum ether); ${ }^{1} \mathrm{H} \mathrm{nmr}$ (deuteriochloroform): $\delta 3.33$ (dd, J = 8, $16.5 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}$ ), 3.82 (dd, J = $9.5,16.5 \mathrm{~Hz}, 1 \mathrm{H}, 3-\mathrm{H}), 5.83(\mathrm{dd}, \mathrm{J}=8,9.5 \mathrm{~Hz}, 1 \mathrm{H}, 2-\mathrm{H}), 6.94(\mathrm{~d}$, $\mathrm{J}=5.5 \mathrm{~Hz}, 1 \mathrm{H}, 5-\mathrm{H}), 7.25-7.53(\mathrm{~m}, 10 \mathrm{H}$, aromatic H$), 8.12 \mathrm{ppm}$ (d, J = $5.5 \mathrm{~Hz}, 1 \mathrm{H}, 6-\mathrm{H}$ ); ms: m/z $274[\mathrm{M}+\mathrm{H}]^{+}$.
Anal. Calcd. for $\mathrm{C}_{19} \mathrm{H}_{15} \mathrm{NO}: \mathrm{C}, 83.49 ; \mathrm{H}, 5.53 ; \mathrm{N}, 5.12$. Found: C, 83.47; H, 5.67; N, 5.11.
The Preparation of $\mathbf{6 d}$ from $\mathbf{5 d}$ and Sodium Methoxide.
A mixture of $\mathbf{5 d}(1.42 \mathrm{~g}, 5 \mathrm{mmoles})$ in a solution of sodium ( $0.23 \mathrm{~g}, 10 \mathrm{mmoles}$ ) in anhydrous methanol ( 50 ml ) was refluxed for 3 hours. After removal of the solvent in vacuo, cold water was added to the residue. The resulting mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and concentrated in vacuo. The residue was purified by column chromatography on silica gel with methylene chloride as the eluent to afford $\mathbf{6 d}(0.84 \mathrm{~g}, 62 \%)$. The melting point and ir spectrum of this compound coincided with those of an authentic sample prepared from $2 \mathbf{d}$ and sodium methoxide.

## The Preparation of $\mathbf{5 d}$ from $\mathbf{6 d}$ and Sodium Ethoxide.

A mixture of $\mathbf{6 d}(1.35 \mathrm{~g}, 5 \mathrm{mmoles})$ in a solution of sodium ( $0.23 \mathrm{~g}, 10 \mathrm{mmoles}$ ) in anhydrous ethanol ( 50 ml ) was refluxed for 3 hours. After work-up as described above, compound 5d ( $0.31 \mathrm{~g}, 22 \%$ ) was obtained. The melting point and ir spectrum of this compound coincided with those of an authentic sample prepared from 2d and sodium ethoxide.

General Procedure for the Preparation of 10a-c and 11a-c from 1a-c and Benzalacetone or Cinnamaldehyde.

A mixture of 1a-c ( 10 mmoles ) and benzalacetone ( $1.75 \mathrm{~g}, 12$ mmoles) or cinnamaldehyde ( $1.58 \mathrm{~g}, 12$ mmoles) in a solution of sodium ( $0.46 \mathrm{~g}, 20 \mathrm{mmoles}$ ) in anhydrous ethanol ( 30 ml ) was stirred at room temperature overnight, and then refluxed for 1 hours. After removal of the solvent in vacuo, cold water was added to the residue. The resulting mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and concentrated in vacuo. The residue was purified by column
chromatography on silica gel with methylene chloride as the eluent to afford $\mathbf{1 0 a}(0.97 \mathrm{~g}, 46 \%), \mathbf{1 0 b}(0.99 \mathrm{~g}, 44 \%), \mathbf{1 0 c}(1.29 \mathrm{~g}$, $45 \%)$, 11a $(0.86 \mathrm{~g}, 44 \%)$, 11b $(0.79 \mathrm{~g}, 37 \%)$ and 11c $(0.83 \mathrm{~g}$, $30 \%$ ), respectively.
General Procedure for the Preparation of 9a-d from 1a-d and Methyl Vinyl Ketone.

A mixture of 1a-d (10 mmoles) and methyl vinyl ketone ( 0.84 $\mathrm{g}, 12 \mathrm{mmoles})$ in a solution of sodium ( $0.46 \mathrm{~g}, 20 \mathrm{mmoles}$ ) in anhydrous ethanol ( 30 ml ) was stirred at room temperature overnight. After removal of the solvent in vacuo, cold water was added to the residue. The resulting mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and concentrated in vacuo. After potassium tert-butoxide $(2.24 \mathrm{~g}$, 20 mmoles) was added to the residue in anhydrous tert-butyl alcohol ( 30 ml ), the mixture was refluxed for 1 hour. After removal of the solvent in vacuo, cold water was added to the residue. The resulting mixture was extracted with chloroform. The extract was dried over anhydrous sodium sulfate and concentrated in vacuo. The residue was purified by column chromatography on silica gel with methylene chloride as the eluent to yield 9a ( $0.58 \mathrm{~g}, 43 \%$ ), 9b $(0.82 \mathrm{~g}, 55 \%), 9 \mathrm{c}(1.01 \mathrm{~g}, 48 \%)$ and 9 d ( $0.96 \mathrm{~g}, 45 \%$ ), respectively.

## REFERENCES AND NOTES

[1] T. Matsuda, K. Yamagata, Y. Tomioka and M. Yamazaki, Chem. Pharm. Bull., 33, 937 (1985).
[2] E. C. Taylor and A. Mckillop, Advances in Organic Chemistry: Methods and Results, Vol 7, E. C. Taylor, ed, Interscience Publishers, Inc., New York, 1970, p 3.
[3] K. Yamagata, Y. Tomioka and M. Yamazaki, Chem. Pharm. Bull., 34, 590 (1986).
[4] G. H. Alt and A. G. Cook, Enamines: Synthesis, Structure and Reactions, 2nd ed., A. G. Cook, ed, Marcel Dekker, Inc., New York and Basel, 1988, p 189.
[5] S. F. Dyke, The Chemistry of Enamines, Cambridge University Press 1973, p 31.
[6] L. A. Paquette, Principles of Modern Heterocyclic Chemistry, W. A. Benjamin, Inc., New York, 1968, p 225.
[7] F. Brody and P. R. Ruby, Pyridine and Its Derivatives, Vol 14, Part 1, E. Klingsberg, ed, Interscience Publishers, Inc., New York, 1960, p 99.
[8] F. Bossert, H. Meyer and E. Wehinger, Angew. Chem., Int. Ed. Engl., 20, 762 (1981).
[9] A. W. Erian, Chem. Rev., 93, 1991 (1993).
[10] J. M. Robinson, L. W. Brent, C. Chau, K. A. Floyd, S. L. Gillham, T. L. McMahan, D. J. Magda, T. J. Motycka, M. J. Pack, A. L. Roberts, L. A. Seally, S. L. Simpson, R. R. Smith and K. N. Zalesny, J. Org. Chem., 57, 7352 (1992).
[11] I. Katsuyama, S. Ogawa, Y. Yamaguchi, K. Funabiki, M. Matsui, H. Muramatsu and K. Shibata, Synthesis, 1321 (1997).
[12] K. Shibata, I. Katsuyama, H. Izoe, M. Matsui and H. Muramatsu, J. Heterocyclic Chem., 30, 277 (1993), and references cited therein.

